



# Poroelectric Effect on Fracture of Gels

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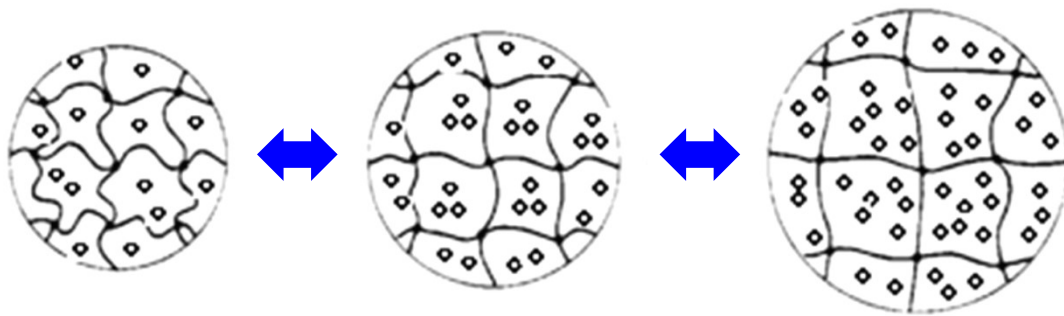
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## Hydrogel is Poroelastic

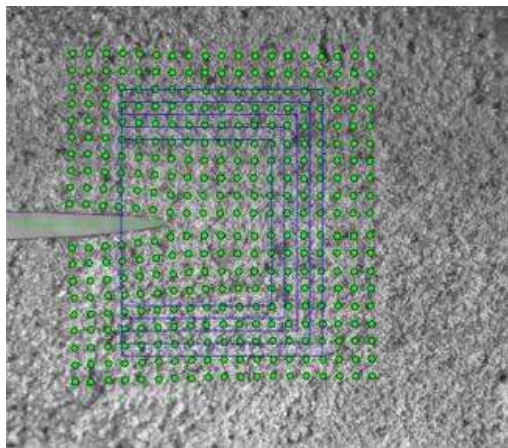


- **Nonlinear Poroelasticity:** large and reversible deformation of polymer network coupled with migration (diffusion) of solvent molecules, in response to mechanical, chemical, and other environmental stimuli (temperature, pH, light, etc)
- Applications: biomedical, soft machines...

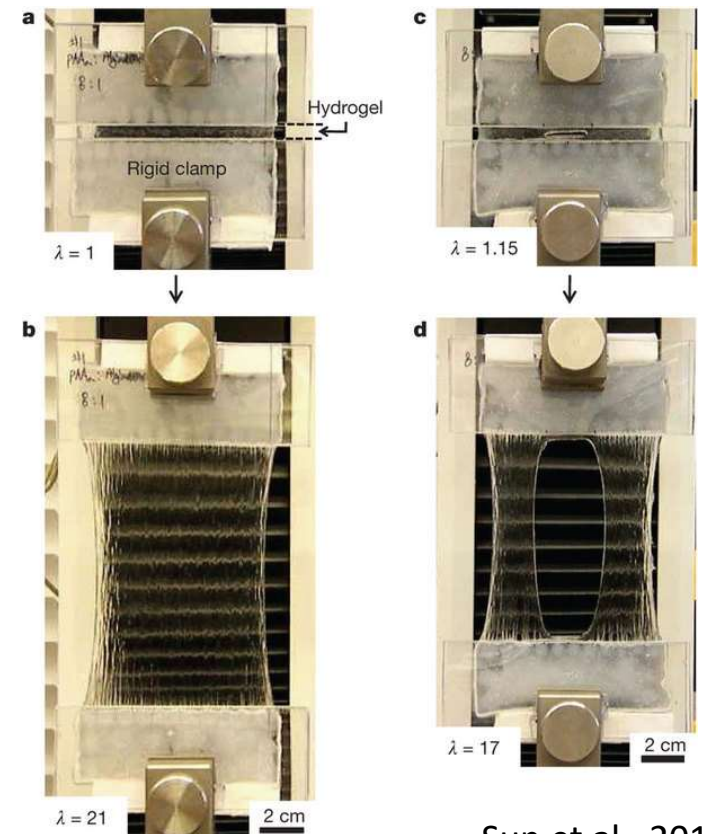


# Fracture Toughness of Gels

- Reported values of fracture toughness range from 1 to  $10^4$  J/m<sup>2</sup>
- Fracture mechanism may vary over different types of gels
- Potential rate dependence may relate to viscoelasticity or solvent diffusion (poroelasticity) or ...



Kwon et al., 2011.



Sun et al., 2012.

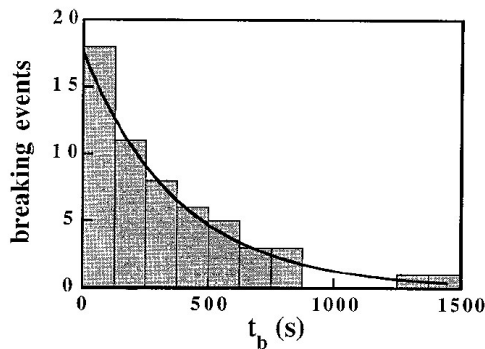
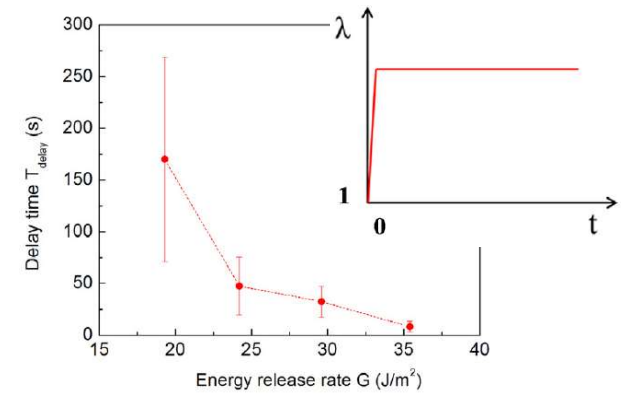
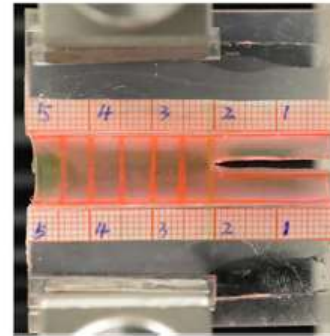


# Delayed Fracture of Gels

Bonn et al., 1998

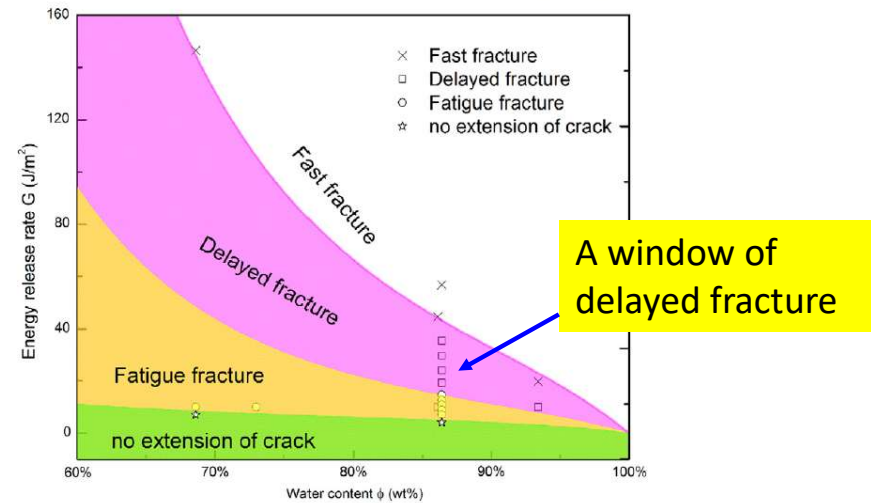


Tang et al., 2017



Time to fracture (nucleation):

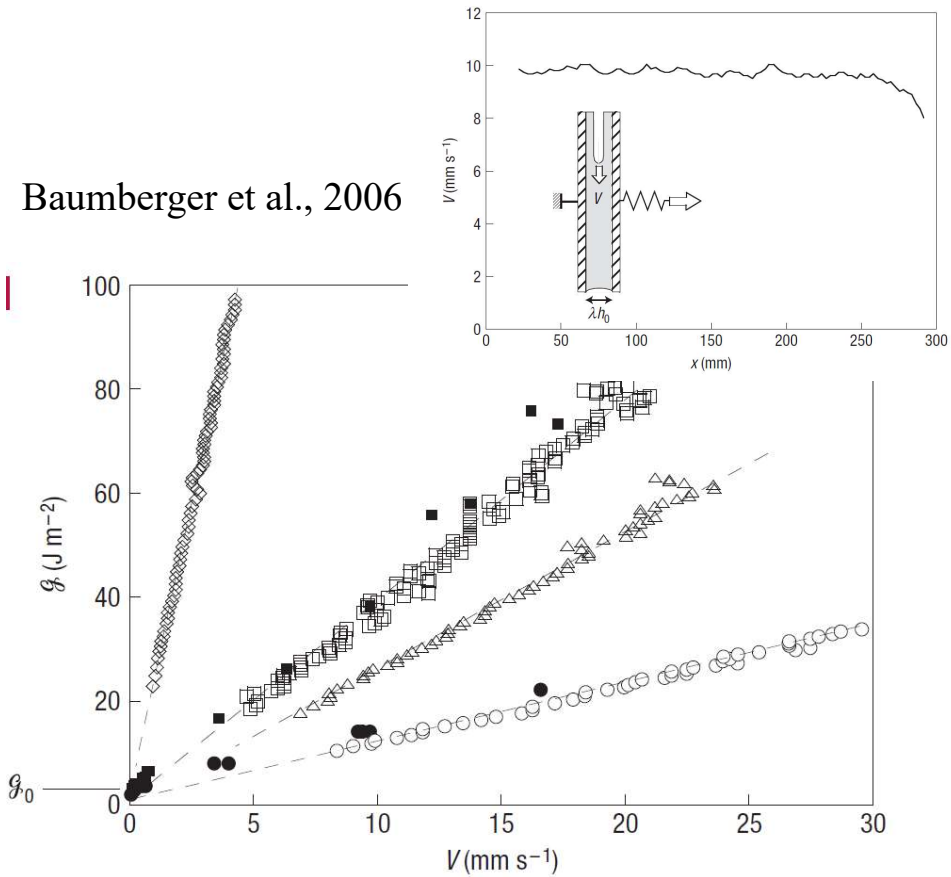
$$\ln t_b \sim \sigma^{-2d+2}$$



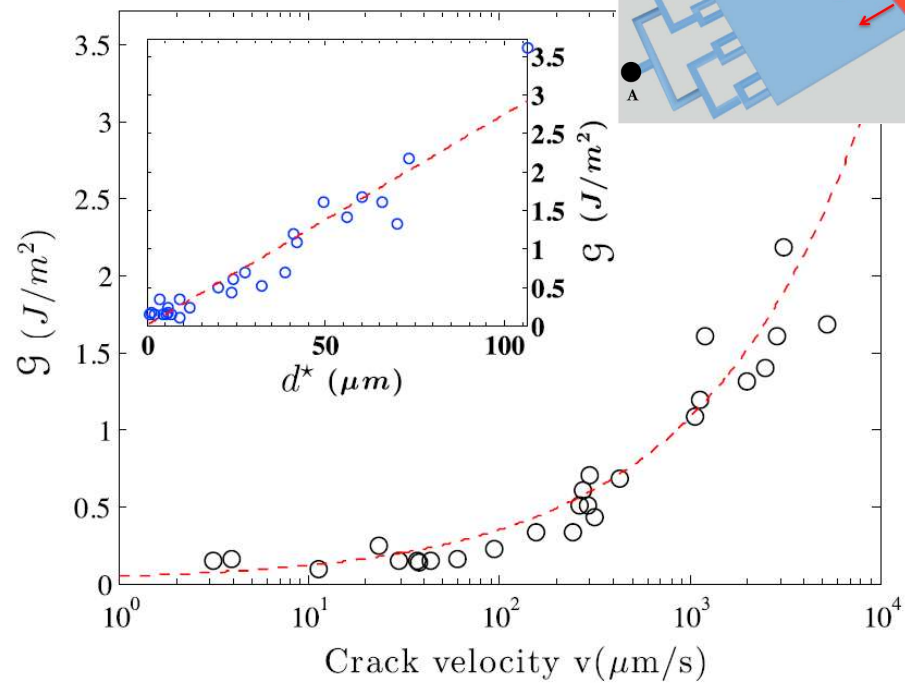


# Rate-dependent Fracture of Gels

Baumberger et al., 2006

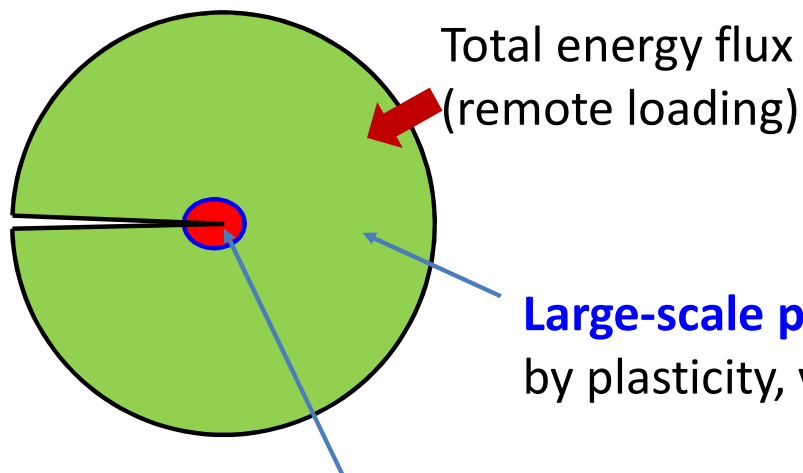


Lefranc and Bouchaud, 2014





## Energetics of Fracture (J-integrals)



$$J_{remote} = J_{elastic} + J_{inelastic}$$

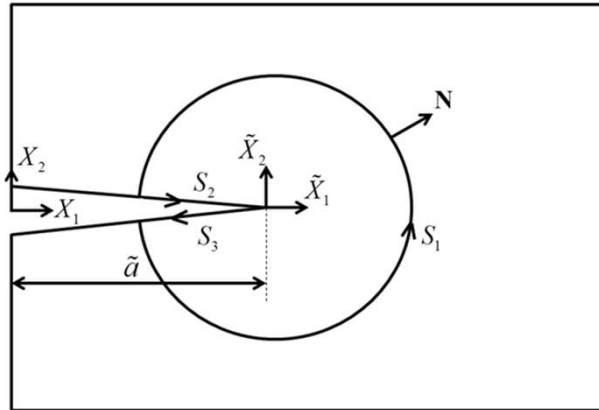
**Small-scale processes**  
(fracture)

$$\Gamma_{fracture} = J_{elastic} = J_{remote} - J_{inelastic}$$

- The elastic energy is stored and (partly) released upon fracture.
- The elastic energy release rate is the driving force for fracture.



## Energy release rate for gels: a modified J-integral



$$J^* = \int_{S_1} \left( UN_1 - s_{iJ} N_J \frac{\partial x_i}{\partial X_1} \right) dS - \int_{V_0} \mu \frac{\partial C}{\partial X_1} dV$$

$$J^* = \int_{S_1} \left( \hat{U} N_1 - s_{iJ} N_J \frac{\partial x_i}{\partial X_1} \right) dS + \int_{V_0} \frac{\partial \mu}{\partial X_1} C dV$$

$$\hat{U}(\mathbf{F}, \mu) = U(\mathbf{F}, C) - \mu C$$

- $J^*$  is path independent.
- The second form is more convenient for numerical calculations.
- A domain integral method can be used to calculate  $J^*$ .



## Proposed Fracture Criteria for Gels

- For a stationary pre-existing crack:

$$J^*(t) = \Gamma_0$$



**Delayed fracture**

- For steady-state crack growth:

$$J^*(\dot{a}) = \Gamma_{SS}$$



**Rate-dependent fracture**

**“Division of labor”:**

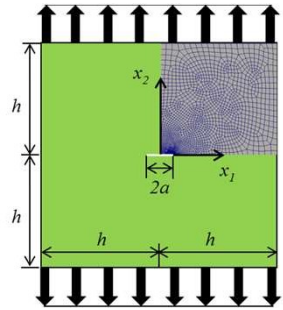
- Calculate the crack-tip energy release rate
- Measure the *intrinsic* fracture toughness



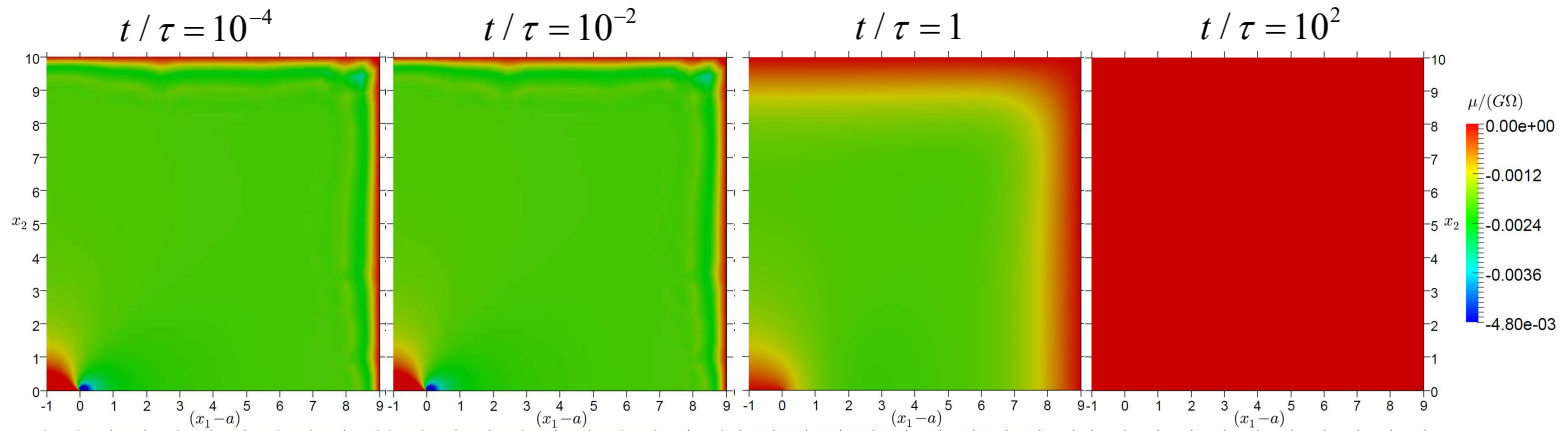


# Transient full-field finite element analysis

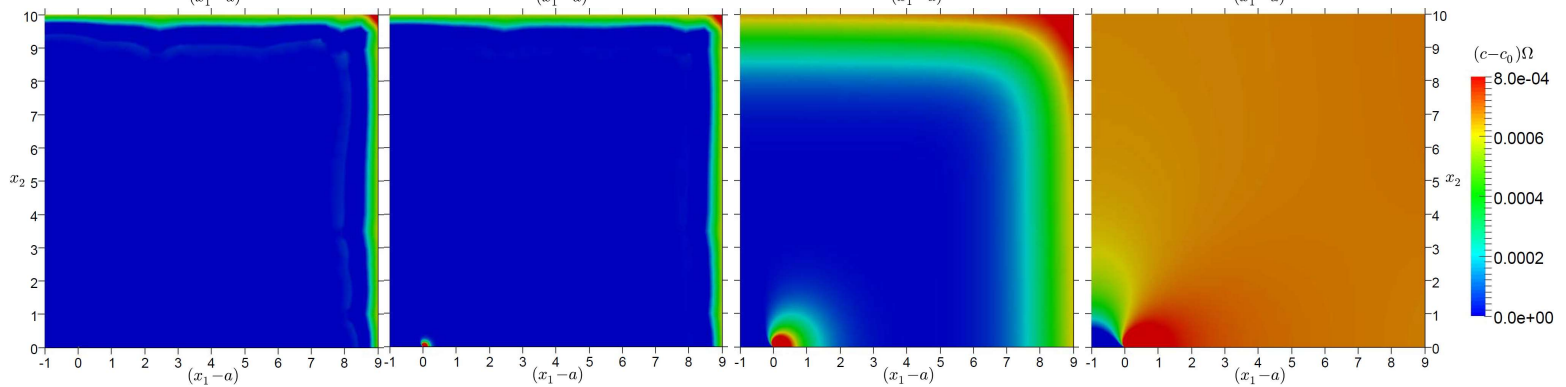
$$\tau = a^2 / D^*$$



Chemical potential

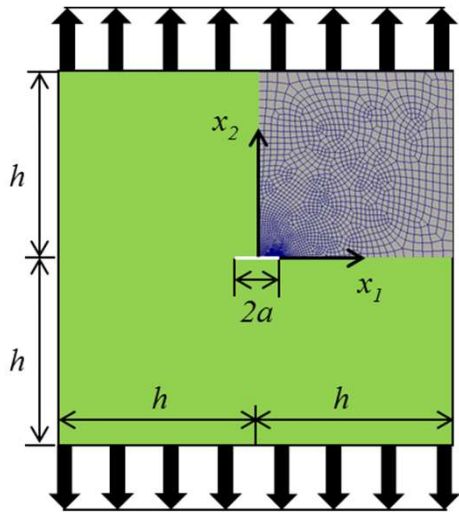


Solvent concentration





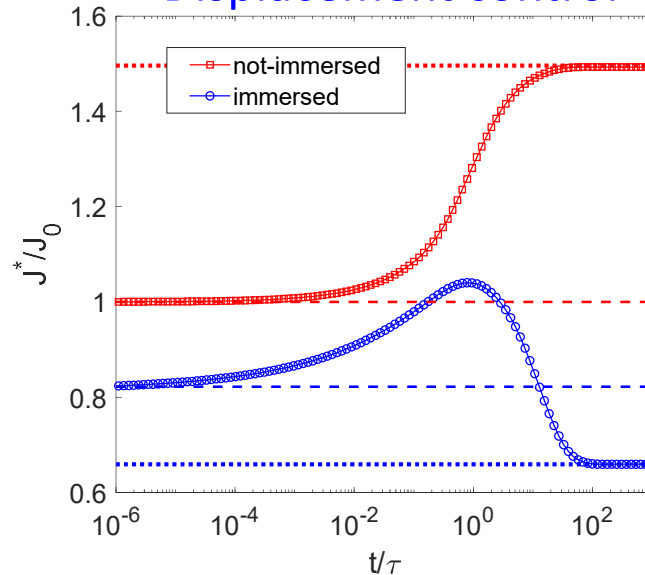
# Time-dependent Energy Release Rate



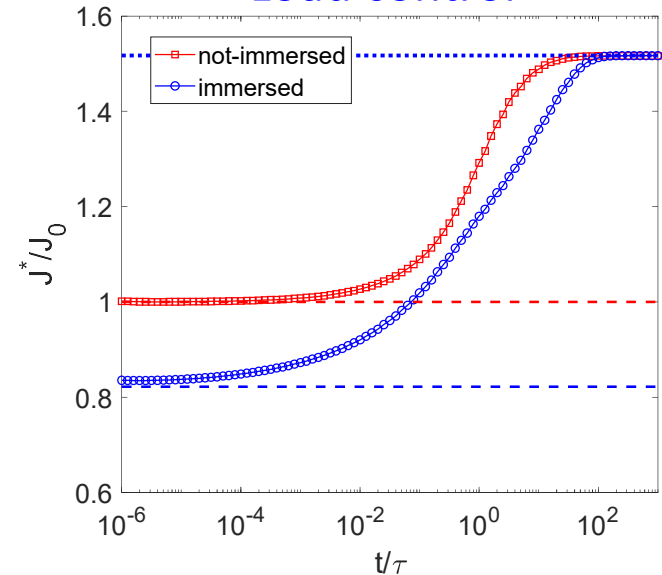
$$\frac{J^*}{J_0} = \Lambda \left( \frac{t}{\tau}, \frac{h}{a}, \nu \right)$$

$$J_0 \sim \sigma_h^2 a / G$$

### Displacement control



### Load control



- The short-time limits are different for immersed and not-immersed specimens.
- The long-time limits are different for immersed specimens under displacement and load control.

Yu, et al., J. Appl. Mech. 85, no.111011 (2018).

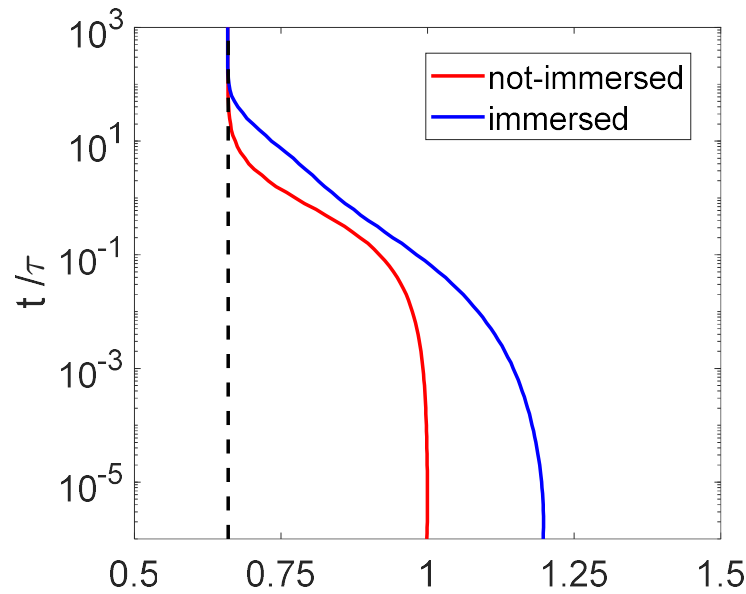
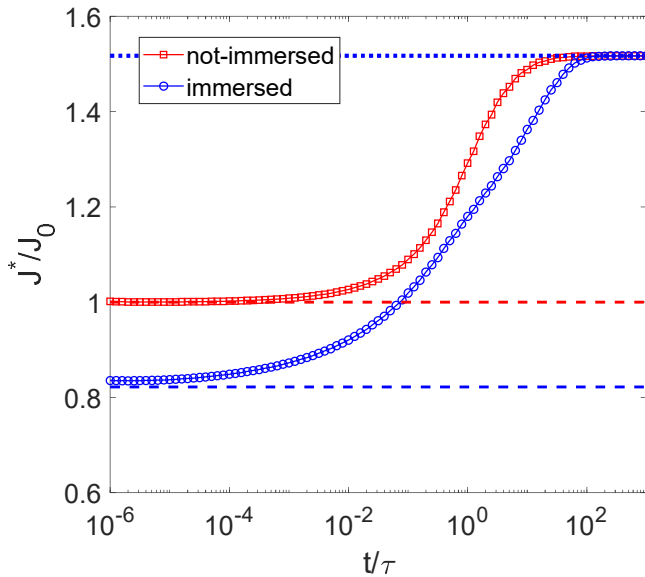


# Delayed Fracture by Poroelasticity

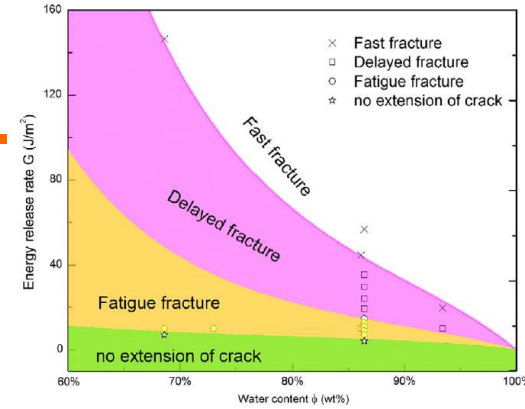
$$J^*(t) = J_0 \Lambda\left(\frac{t}{\tau}\right) = \Gamma$$



$$t_b = \frac{a^2}{D^*} f\left(\frac{J_0}{\Gamma}, \nu\right)$$



$$J_0 \sim \sigma_h^2 a / G \quad J_0 / \Gamma$$



Critical condition for instantaneous fracture:

$$J^*(t \rightarrow 0^+) > \Gamma$$

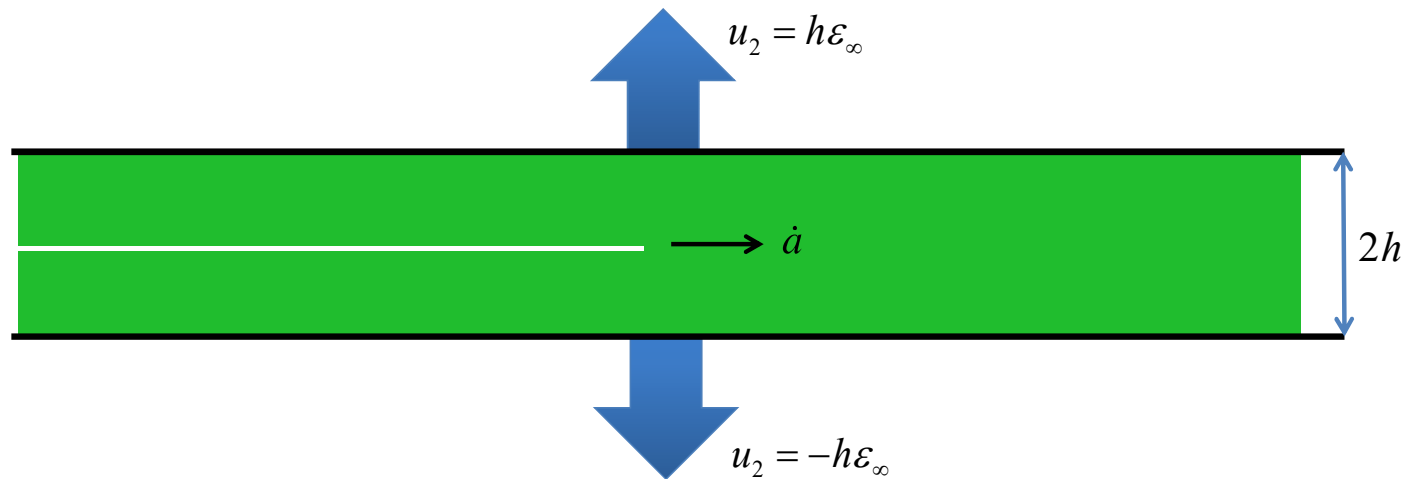
Threshold for delayed fracture:

$$J^*(t \rightarrow \infty) > \Gamma$$

Same threshold but different critical loads for immersed and not-immersed specimens.



## Steady-state crack growth model



*Peclet number*

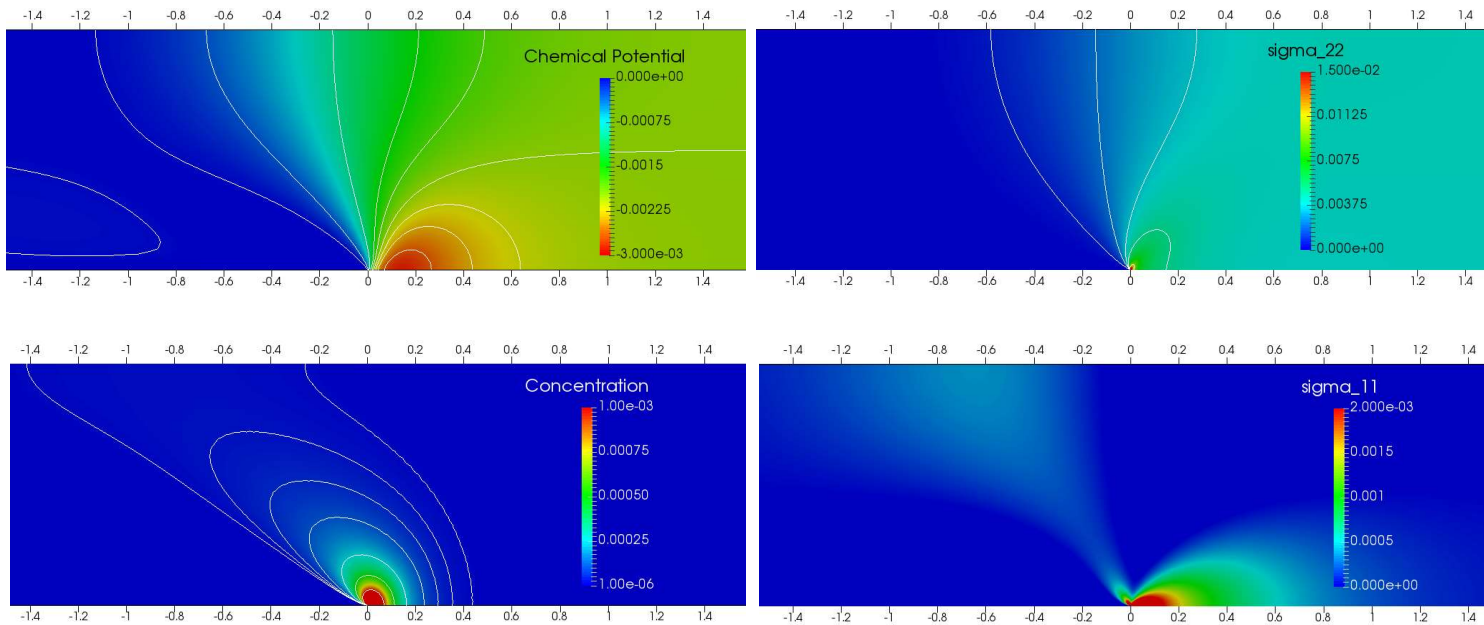
$$Pe = \frac{h}{l_{SS}} = \frac{\dot{a}h}{D^*}$$

- A semi-infinite crack in an infinitely long strip made of a **linearly poroelastic** material;
- Uniform vertical displacements are applied at the top and bottom, while the crack grows in a **steady state**.
- Ignore inertia for quasi-static crack growth.



# Numerical Results

$$Pe = ah / D^* = 10, \nu = 0.2414$$



**Poroeelastic shielding:** the crack-tip stress intensity factor is lower than the elastic case.

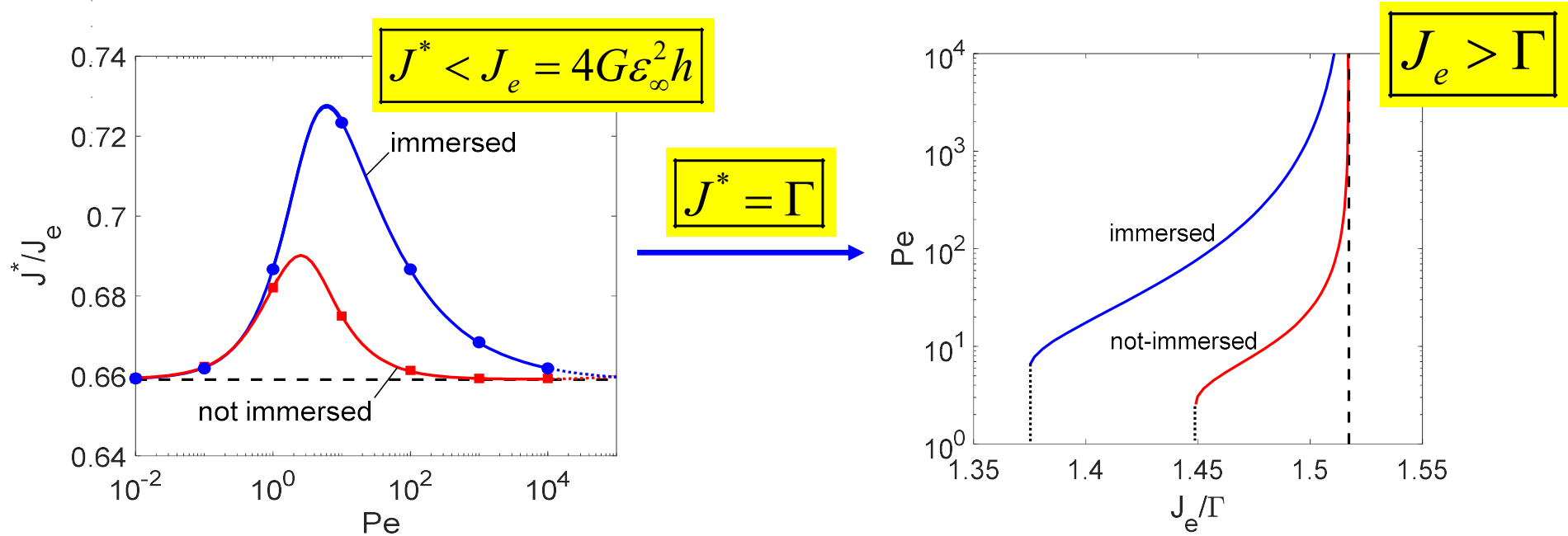
$$\frac{K_{tip}}{K_e} = f(Pe, \nu)$$

$$K_{tip} < K_e = 4G\varepsilon_\infty \sqrt{h}$$

Yu, et al., J. Mech. Phys. Solids 118, 15-39 (2018).



# Poroelastic Toughening



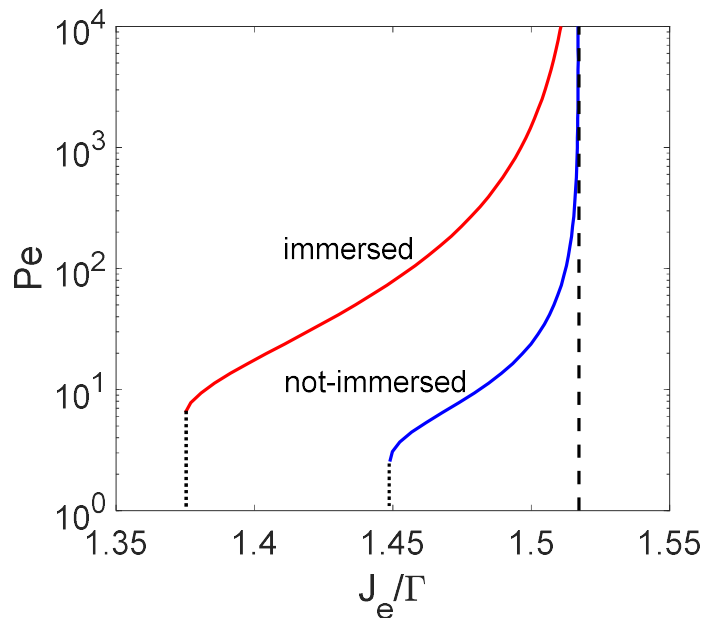
The apparent energy release rate is greater than the intrinsic fracture energy (toughness) because energy dissipation by solvent diffusion around the crack tip.



## Experimental implications

Measure intrinsic fracture toughness as a function of crack speed:

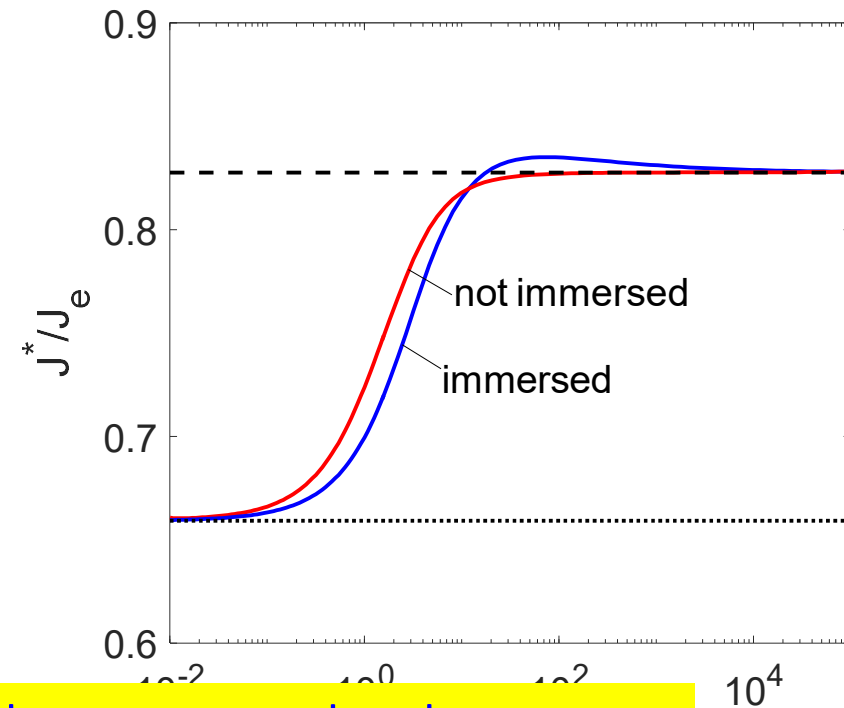
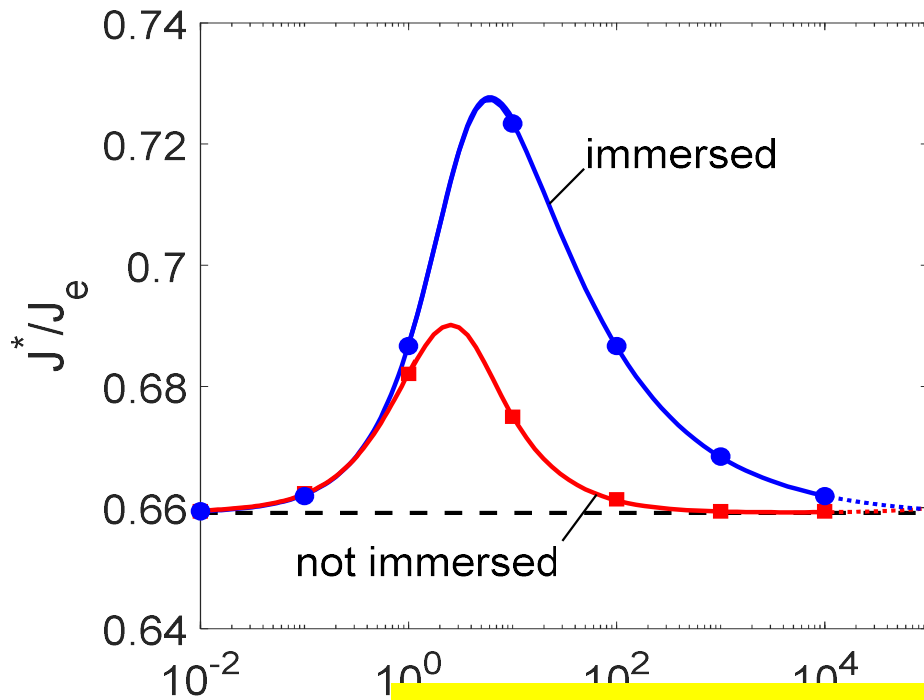
$$\Gamma = J^* = 4G\varepsilon_\infty^2 h \Lambda \left( \frac{\dot{a}h}{D^*}, \nu \right)$$



- “velocity toughening”: the apparent fracture energy increases with crack speed.
- Effect of solvent viscosity: high viscosity  $\rightarrow$  low diffusivity  $\rightarrow$  low crack speed
- Effect of crack-tip soaking: immersed (wet) versus not-immersed (dry)



## Plane Strain versus Plane Stress

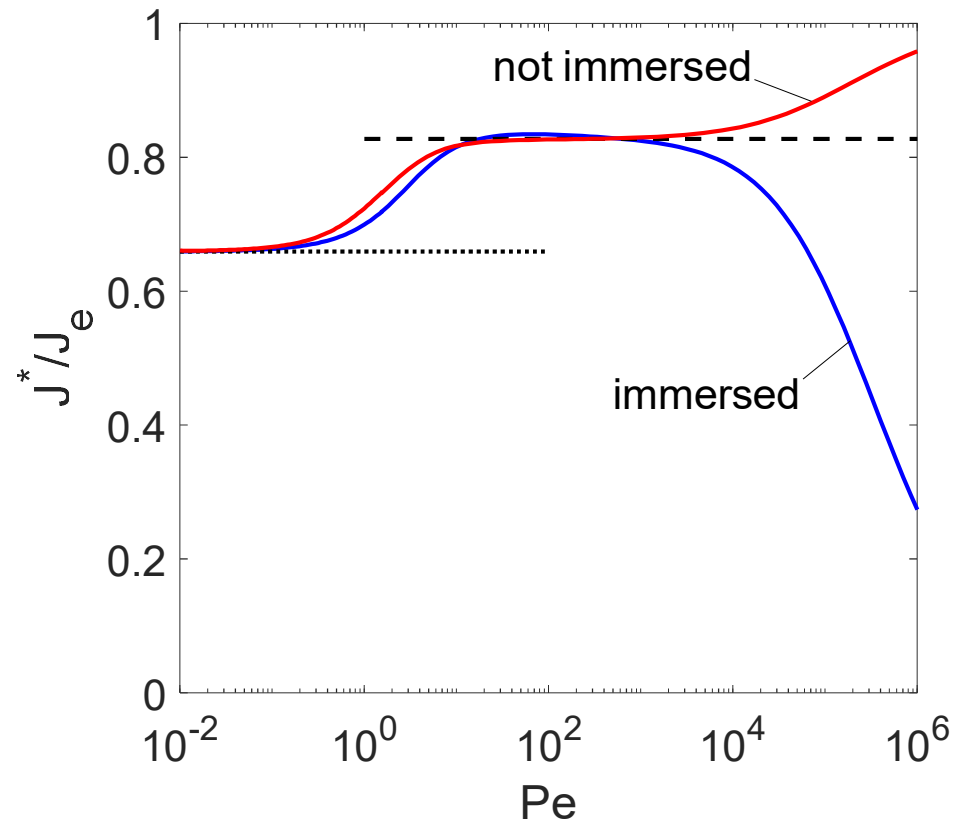
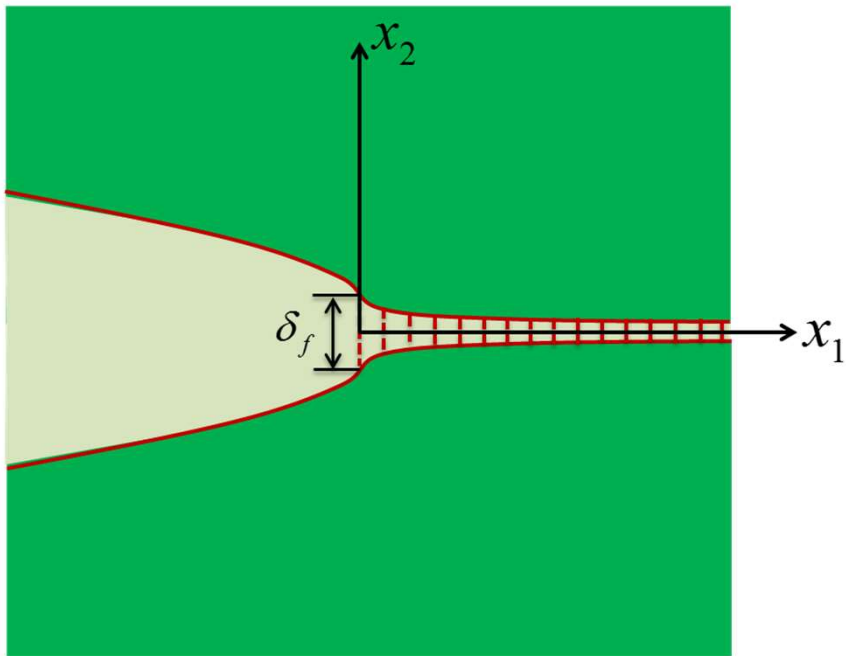


Solvent diffusion dissipates less energy under plane stress, opposite to the plastic dissipation in metals.





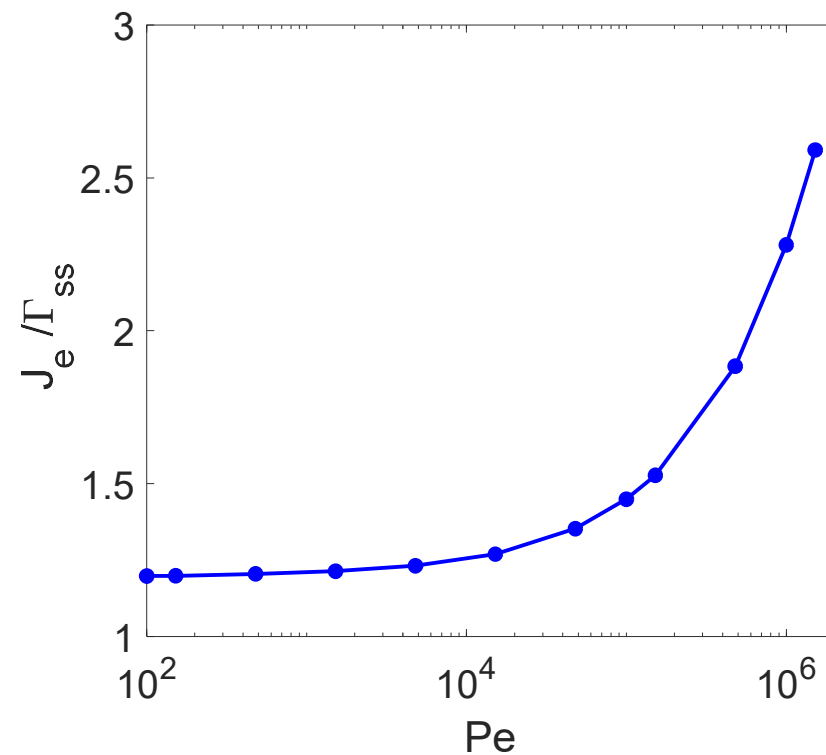
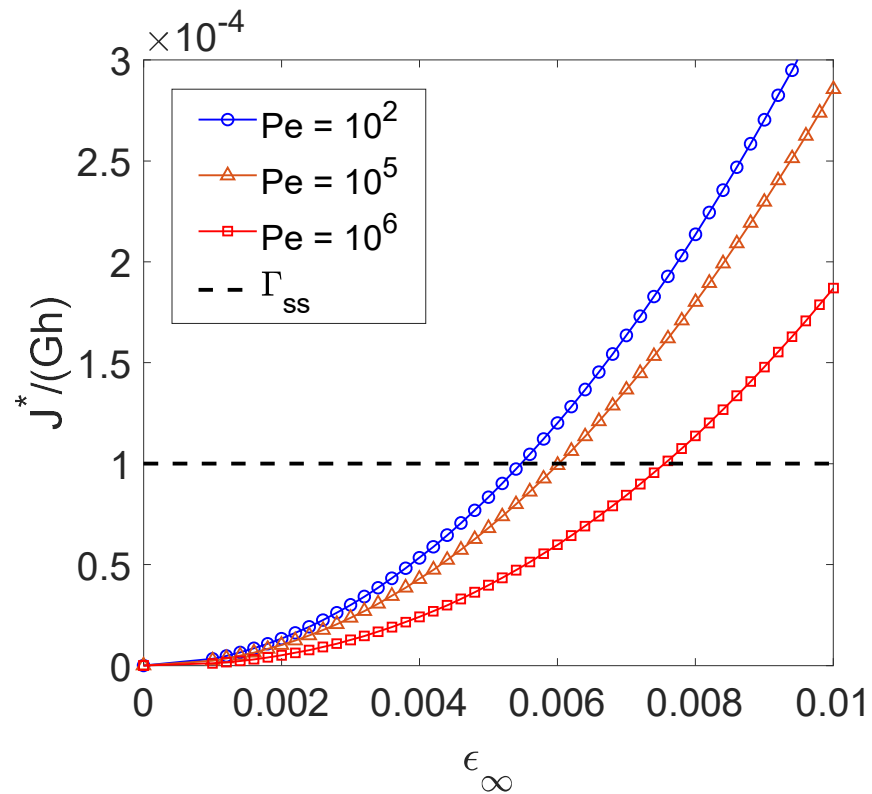
# A poroelastic cohesive zone model



Yu, et al., J. Appl. Mech. (2019).



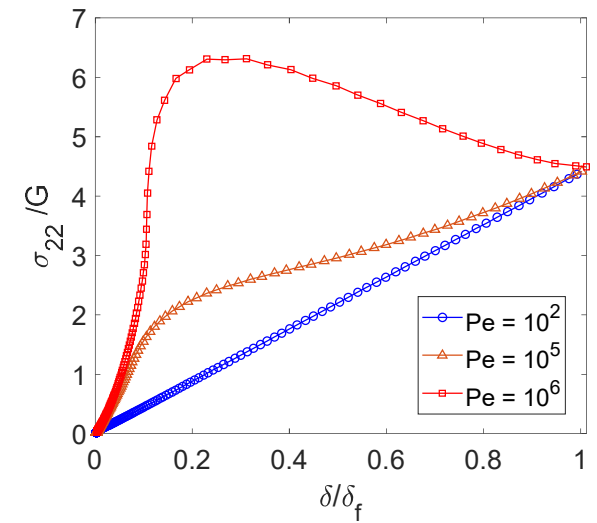
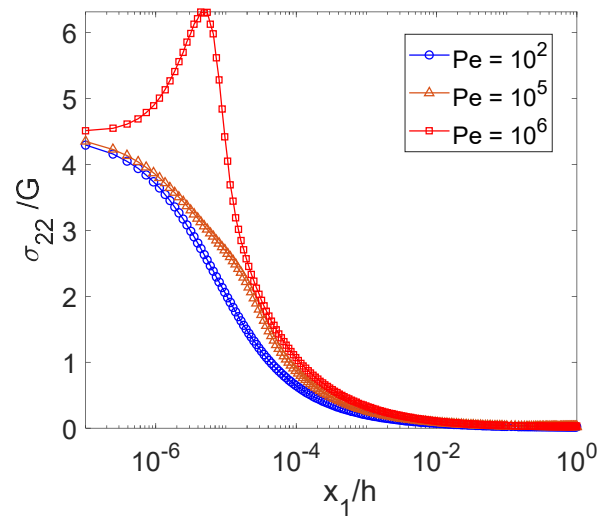
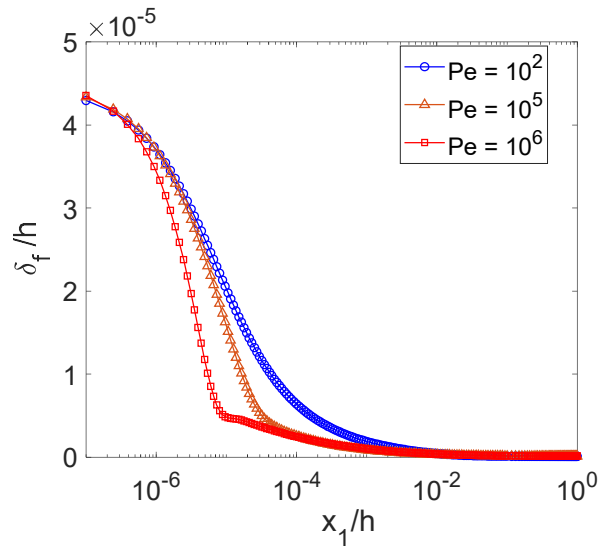
# Rate-dependent fracture toughness



Yu, et al., J. Appl. Mech. (2019).



# Rate-dependent traction-separation



Solvent diffusion within the cohesive introduces additional toughening effect, leading to rate-dependent fracture.

Yu, et al., J. Appl. Mech. (2019).



## Summary: Fracture Criteria for Gels

- For a stationary pre-existing crack:

$$J^*(t) = \Gamma_0$$



**Delayed fracture**

- For steady-state crack growth:

$$J^*(\dot{a}) = \Gamma_{SS}$$



**Rate-dependent fracture**

- A poroelastic cohesive zone model predicts rate-dependent traction-separation relations and additional toughening.